



Remediation Technologies and Requalification of the Territory

Practical Applications for Nanoremediation

Session 1

RemTech, Ferrara Exhibition Center, Ferrara, Italy

21 September 2016







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Agenda for Session 1

Time (Hrs)	Title	Presenter
0930 - 1030	A Primer on Nanoremediation – History, Applications, and Issues	D.W. Elliott
1030 - 1045	Break	
1045 - 1145	Nanoremediation in the EU - Impacts of NanoRem and Technology Combinations	M. Cernik
1145 - 1200	Break	
1200 - 1300	Key Field Applications of Nanoremediation – Lessons Learned and Future Directions	P. Kvapil

Technical university of Liberec





1100 employes 9000 students 6 faculties + 2 institutesInstitute for Nanomaterials, Advanced Technologies and



Innovation

Nanoremediation in EU

- History
 - 2004: 1st application outside of USA
 - 2005: pilot test at Kurovody
 - Overview of Czech sites
- Characterization of the nanoparticles
 - Different techniques for nZVI characterization
 - Slurry, Dry particles, Milled particles
 - Reactivity with samples of contaminated water
- Novel deployment approaches
 - Combination of nZVI and bioremediation (ARD)
 - Electrokinetics (EK) and nZVI

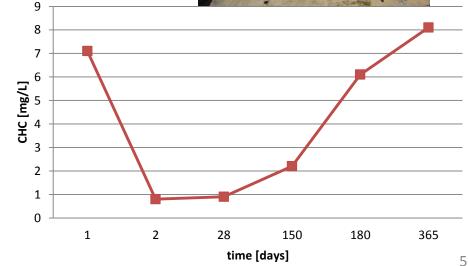
History of nZVI in Europe

1st application in the Czech Republic

- 2004 1st pilot test (Spolchemie, with Golder) based on information from M. Pupeza
- Results:
 - ORP decrease (from +100 to -500 mV)
 - pH increase (from 6.5 to 8.5)
 - CHC decrease for 6 months







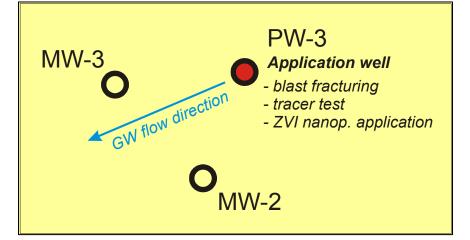
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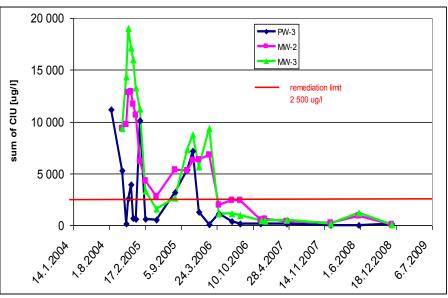
Czech nZVI experience

2nd nZVI application

- 2005 fractured bedrock (Kurivody, CZ)
- Tracer test
- Pilot test
- No rebound effect









Czech nZVI experience

Site	Cont.	Lab (L)/Pilot(P)/ Remediation (R)	nZVI
Spolchemie 2004	CI-Ethenes	L,P – 20 kg	ZHANG,
Kuřívody 2005, 2006	CI-Ethenes	L,P – 50 kg	ZHANG, RNIP
Piešťany 2005	CI-Ethenes	L,P – 20 kg	ZHANG
Permon 2006	Cr6+	L,P – 7 kg	RNIP
Rožmitál 2007-9	РСВ	L,P – 150 kg	RNIP, NANOFER
Hluk 2007, 2008 (PRB)	CI-Ethenes	L,P – 150 kg	RNIP, NANOFER
Hořice 2008, 2009-2013	CI-Ethenes	L,P,R – 600 kg	RNIP, NANOFER, DC, BIO
Uherský Brod 2008	CI-Ethenes	P – 50 kg	NANOFER
Písečná 2008-2011, 2014-17	CI-E, CI-A	L,P,R – 250 kg	RNIP, NANOFER
Uzin 2009	CI-E	L,P – 150 kg	NANOFER
Spolchemie 2009-2017	CI-E	L,P,R – 1000 kg	NANOFER, DC
MARS 2012-2015	CI-E	L,P,R – 1000 kg	NANOFER, DC, BIO

History of Czech nZVI scientific actions

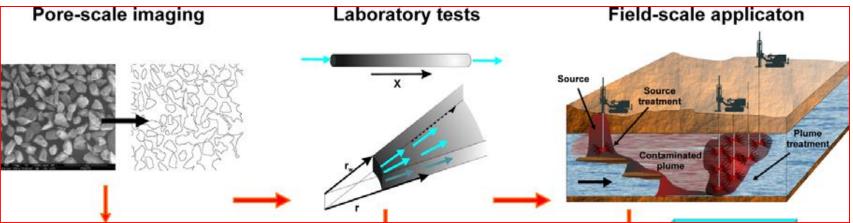
Why just in the Czech Republic?

- No eco-friendly production → changes of political system → privatization → commitment of remediation
- Pump & treat, venting not effective enough
- Authorities support innovative technologies
- Czech project (2006-8): "Development of nZVI production and application for treatment of contaminated groundwater"
- Cooperation of TUL+ UPOL + AQUATEST consultancy
 → NANOIRON establishment
- Pilot and full scale applications, new products development
- Observatory Nano EU project → EU project NANOREM



Taking Nanotechnological **ORem** Remediation Processes from Lab Scale to End User Applications for the **Restoration of a Clean Environment**

- EU project NANOREM
- 29 EU institutions from 15 Countries
- PAG: D. Elliott, G. Lowry, M. Wiesner
- Budget €12 million (\$16.8 million); duration 48 months
- Aim: Identification of the most appropriate nanoremediation technological approaches to achieve a step change in practical remediation performance



List of activities (WPs)

& their objectives



- Design, Improvement and Optimized Production of:
 - Zero-Valent Iron Nanoparticles (WP2)
 - Non-ZVI and Composite Nanoparticles (WP3)
- Mobility and Fate of Nanoparticles (WP4)
- Environmental Impact of Reactive Nanoparticles (WP5)
- Analytical Methods for In-situ Determination of Nanoparticles Fate (WP6)
- Upscaling, Risk and Sustainability (WP8)
- Pilot Site Applications and Field Demonstrations (WP10)

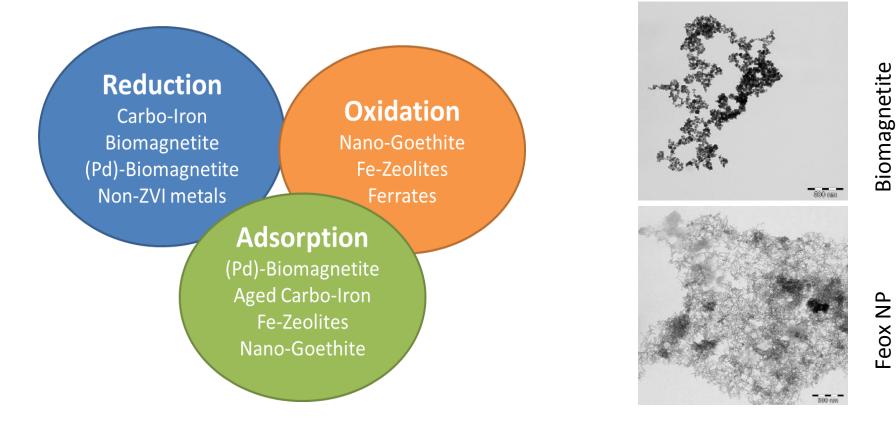






Taking Nanotechnological **NORE** Remediation Processes from Lab Scale to End User Applications for the **Restoration of a Clean Environment**

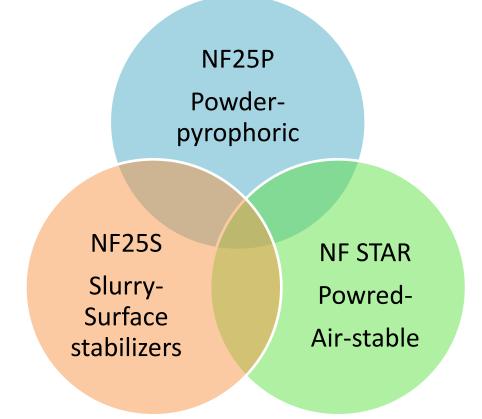
WP3: Non-Fe and combined particles

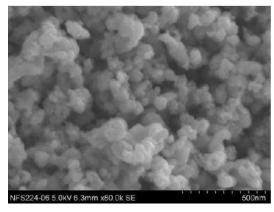


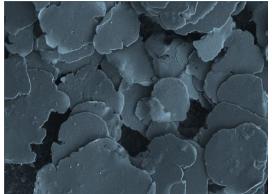


Taking Nanotechnological Remediation Processes from Lab Scale to End User Applications for the Restoration of a Clean Environment

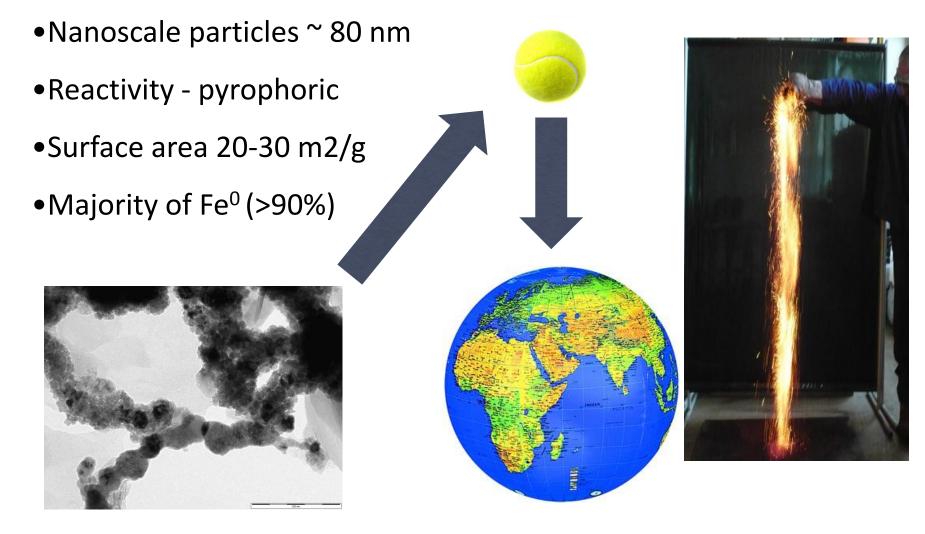
WP2: nZVI particles



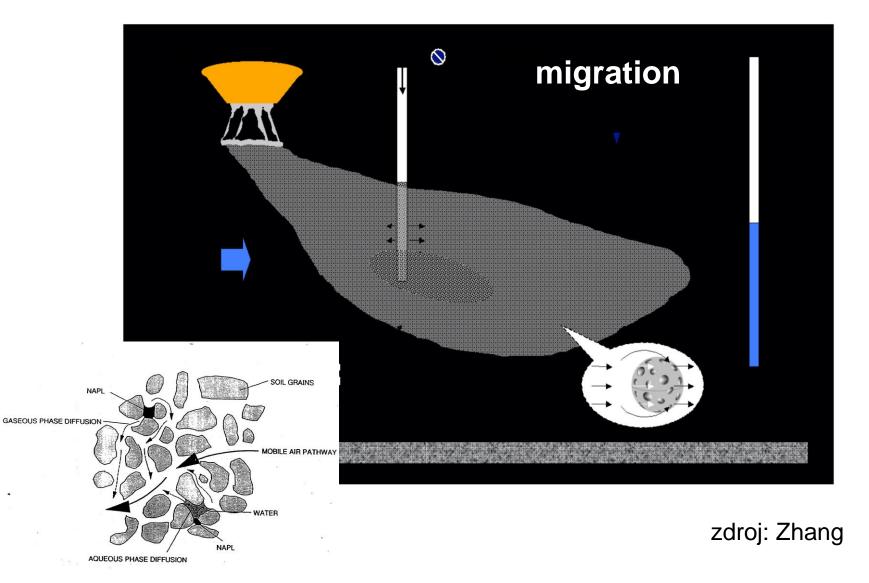




Basic nZVI characteristics



Basic of nanoremediation



Optimal properties

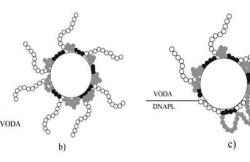
- reactivity with contam.
- mobility in the aquifer
- stability before appl.
- NO (minimum) of negative environ. effects
- price, availability



Inhicor-T Starch Carboxymethyle cellulose Polyacrylic acid Cellulose Tween 60

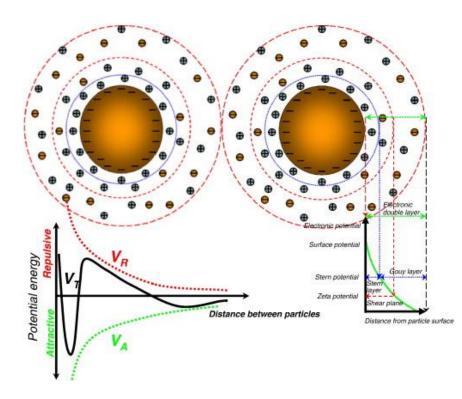


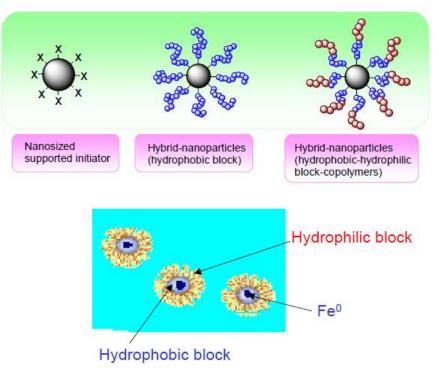




Princip of stabilization

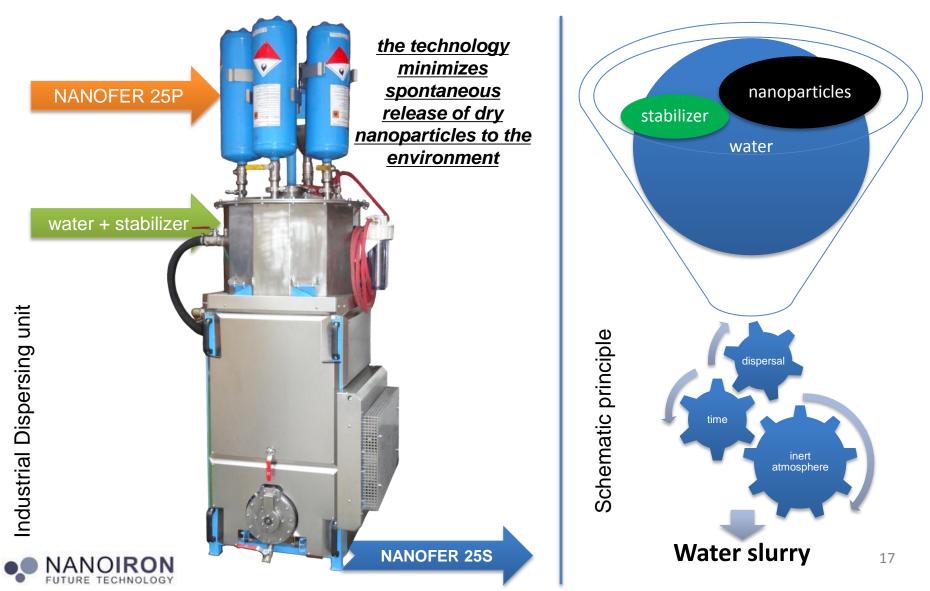
- Electrostatic: increase of surface charge \rightarrow repulsion
- Sterical \rightarrow increase of nZVI distance





nZVI Slurry Manufacturing

Dispersing process: preparation of fresh nZVI slurry

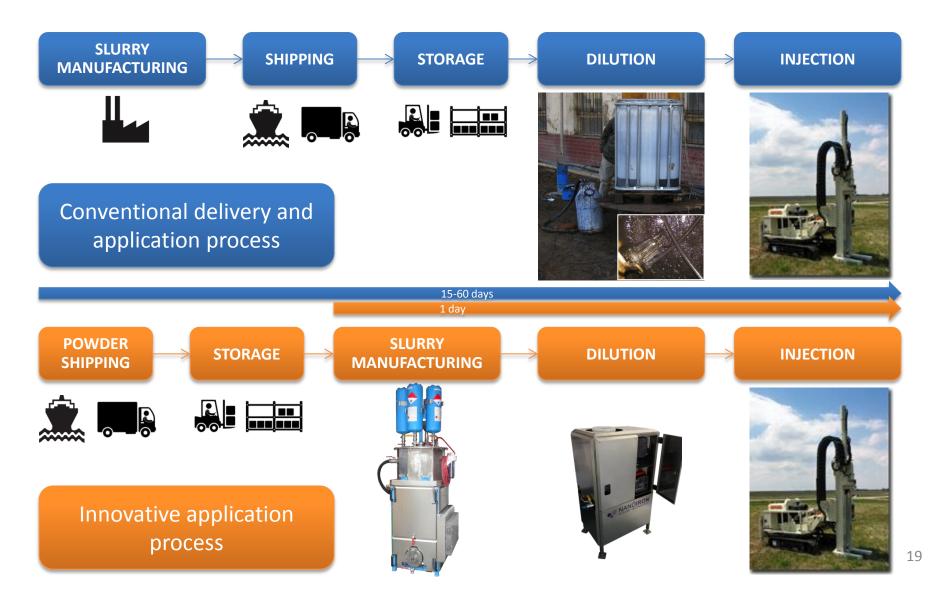


NANOFER STAR

Dry air-stable powder

Innovative nZVI Manufacturing

Field deployment approaches



NANOFER STAR- dry nZVI

Dry powder transported to the site
 On site surface activation, stabilization
 Dillution to a final concentration
 advantage → high reactivity, >95% Fe⁰



Innovative nZVI Manufacturing

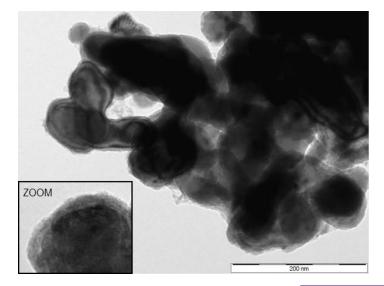
NANOFER STAR: air-stable powder

- Composition (wt%):
 - Min 80% of nZVI
 - Max 20% of iron oxides
- Primary stabilization technique



- STAR means:
 - **Surface stabilized**
 - <u>T</u>ransportable
 - <u>A</u>ir-stable
 - <u>R</u>eactive





No degradation (unlimited storage time)

High reactivity (comparable to NANOFER 25P)

Even cheaper shipping (comparing to NANOFER 25P)

Air stability

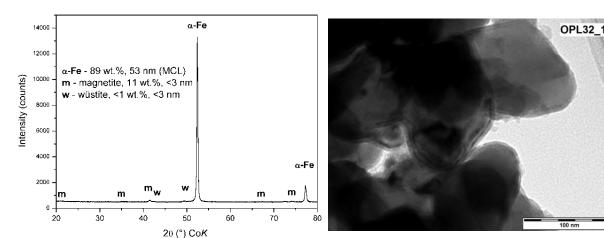
Dangerous classified (UN 3089 - flammable)

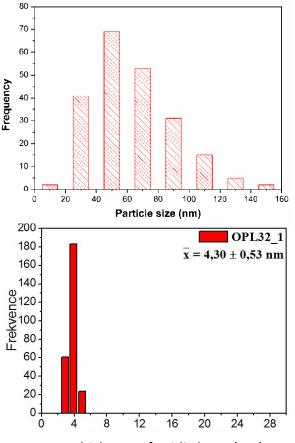
Lower amount of nZVI (comparing to NANOFER 25P)

Challenging nZVI materials

Dry NP powder

- NANOFER STAR dry NP powder produced by solidstate thermal reduction of Fe-oxide
- Tiny oxidic layer for NP protection
- Good stability, transpotability
- Sufficient reactivity (activation)
- Mobility (surface modifications)





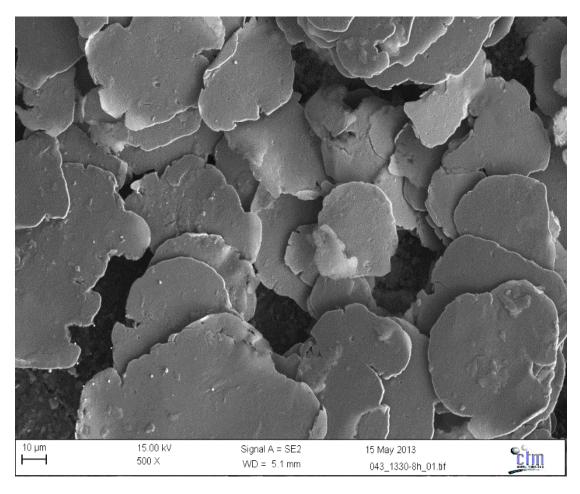
Milling process

Alternative nZVI preparation

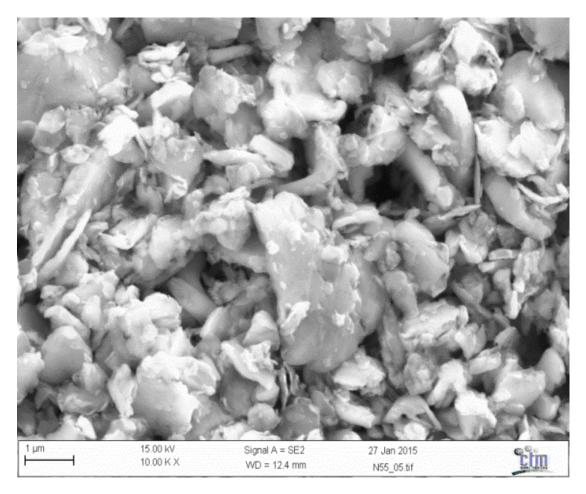


ZVI milling from micro scale

- Preparation of "nano" ZVI particles
- Water or ethanol \rightarrow flakes (< 100 nm thickness)

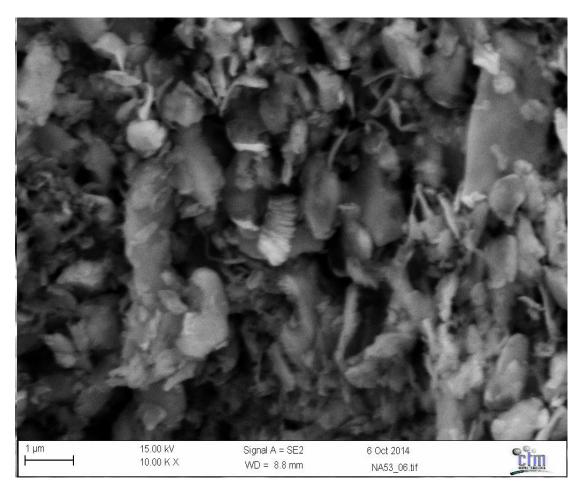


- Preparation of "nano" ZVI particles
- MEG solution



original

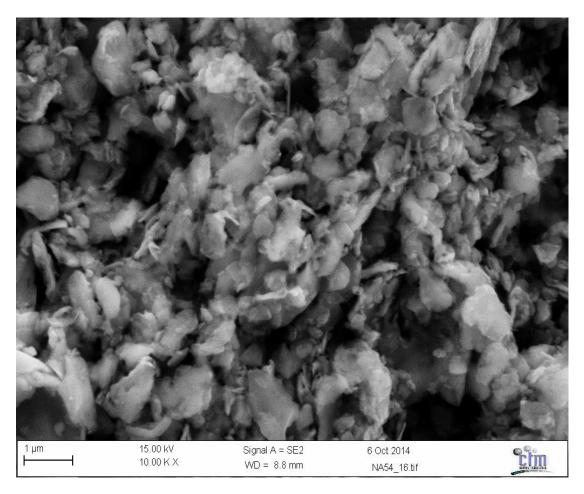
- Preparation of "nano" ZVI particles
- MEG solution



1 g alumina

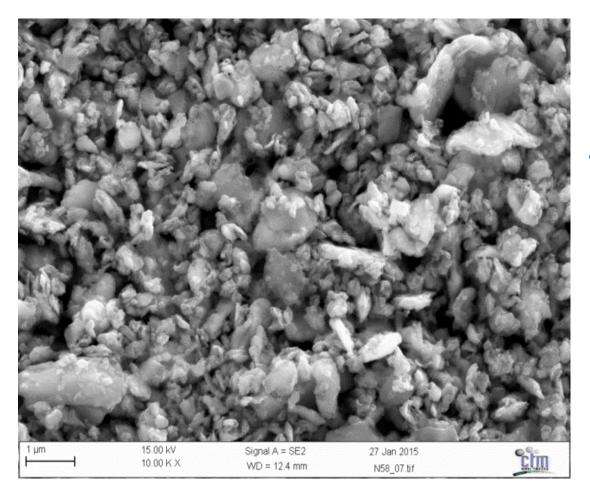


- Preparation of "nano" ZVI particles
- MEG solution



3 g alumina

- Preparation of "nano" ZVI particles
- MEG solution



10 g alumina

400 nm (med) Size < 1 μm (70% volume)

Spherical NP

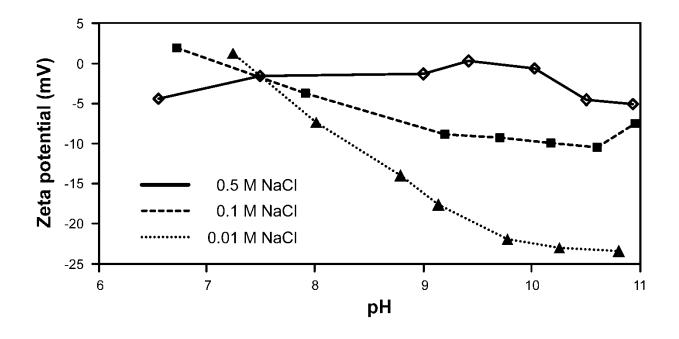
nZVI characterization

3 types of tests

- Structural characteristics:
 - zeta-potential,
 - BET,
 - TEM & SEM, XRD & Mössbauer,
 - Size distribution: DLS & DSC,...
- Reactivity tests:
 - Water (production of H₂ and OH⁻)
 - Selected contaminants (spiked in water)
 - Contaminated water
- Mobility tests
 - 1-D simple tests for comparison
 - Complex 1-D tests
 - 2-D and 3-D tests

ZVI characteristics

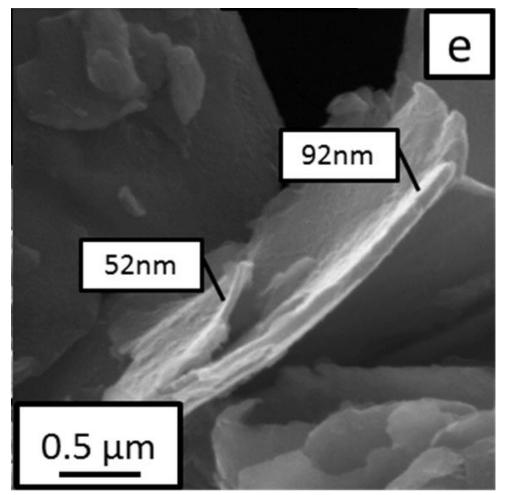
• Zeta-potential



- Addition of NaOH \rightarrow pH increase
- PZC
- Sign and size at gw pH

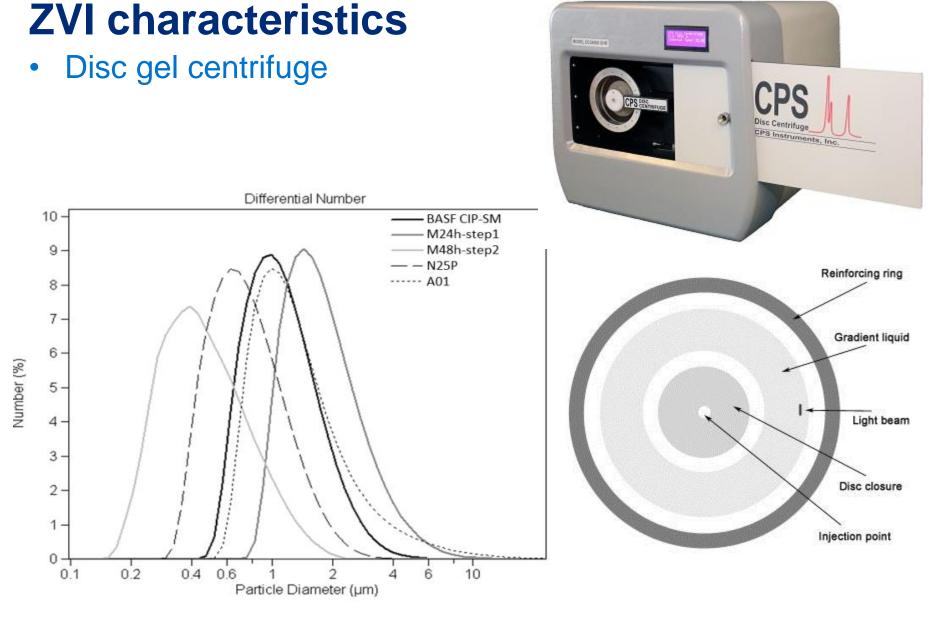
nZVI characteristics

• SEM and TEM pictures



SEM of nZVI after 48h of milling

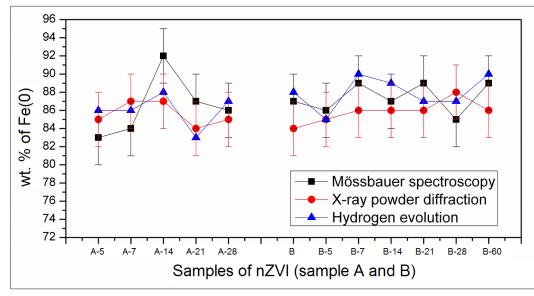




Simple method for reactivity check

QA/QC: Hydrogen evolution method

- Simple, cheap and fast measurement of nZVI content.
- Principle: measurement of hydrogen volume, which evolves due to chemical reaction of nZVI and an acid.
- The method has been compared to Mössbauer spectroscopy and X-ray powder diffraction.

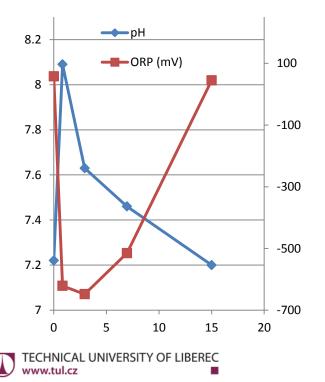


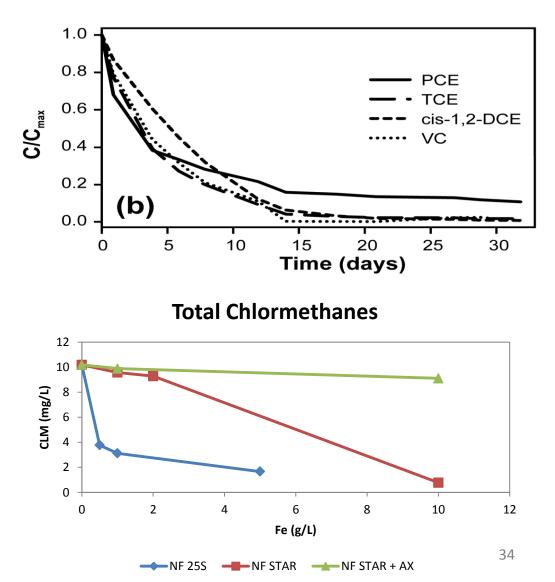


nZVI reactivity

Reaction with contaminated water

- pH and ORP
- Removal of CHC
- Kinetic tests
- Concentration tests

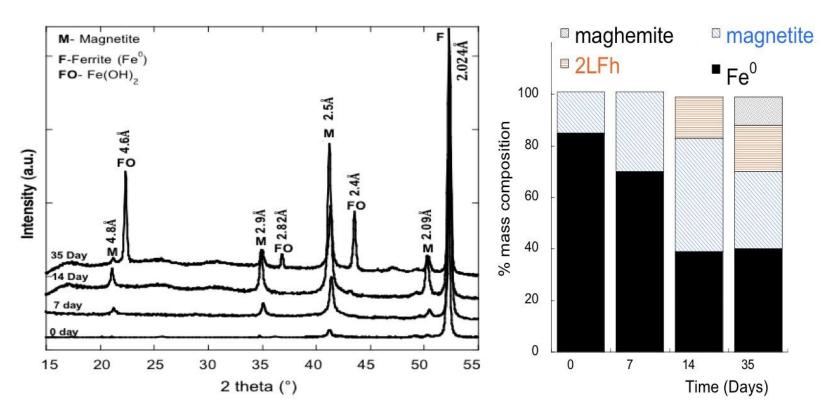




nZVI reactivity

Reaction with distilled water

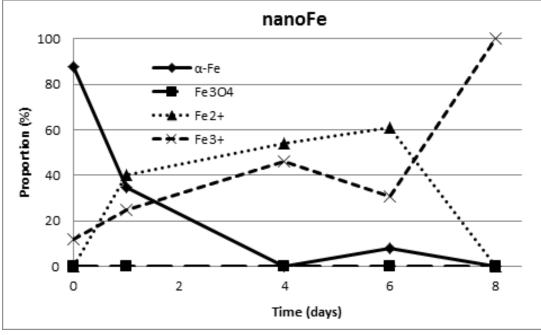
- Reaction in nitrogen atmosphere
- XRD diffractogram → mass composition of nZVI



nZVI reactivity

Reaction with contaminated water

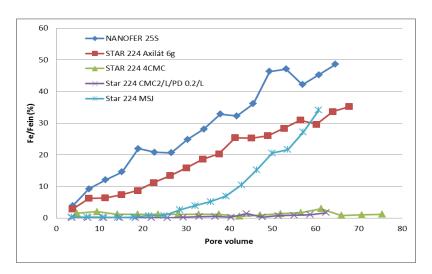
- CHC contaminated water
- Changes in Fe oxidation state composition in time
- Limited nZVI reactivity
- Fe(III)ox final product

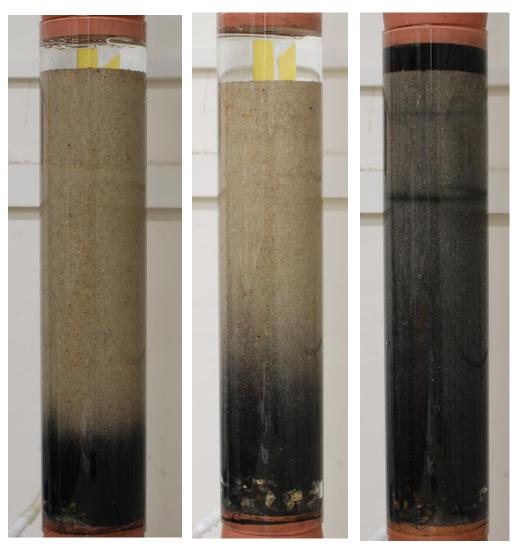


Migration tests

1-D laboratory columns

- Migration bottom- up
- Sandy media
- Low nZVI conc. (<1 g/l)</p>
- Comparison of different modifications







non modified NZVI

modifications

Migration tests

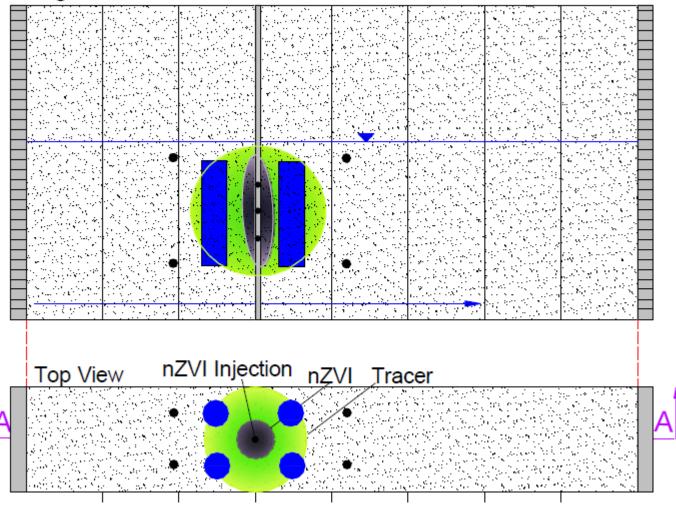
2-D laboratory columns VEGAS Germany (60l, 10g/l, 7 hours)



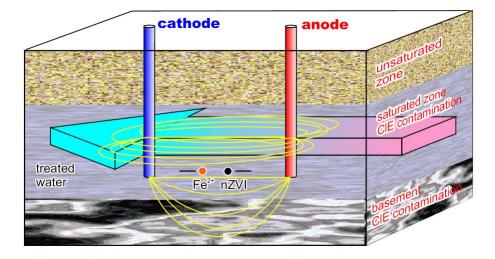
VEGAS – large flume test

600 x 300 x 100 cm Vessel

Longitudinal Section:A-A



Combined methods



Combination of nZVI with other methods

Remediation "trains"

- Why?
- nZVI has limitations
 - High cost (100 €/kg or \$65/lb)
 - Limited migration
 - Low hydraulic conductivity
- Bioremediation has limitations
 - Accumulation of daughter unless bioaugmented (e.g. c-DCE from TCE)
 - Lower ORP needed for dechloration
- Combination of nZVI & other methods
 - with anaerobic biostimulation or bioaugmentation
 - with electrokinetics (DC field)





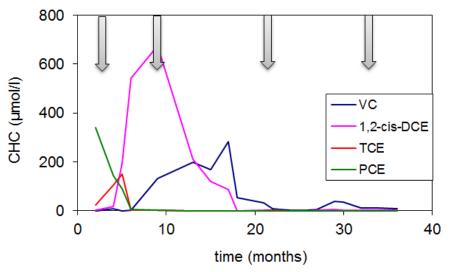
Combination nZVI with bioremediation

Lactate (biostimulation)

- Lactate → fermentation (CO₂ + CH₄) → source of electrons for anaerobic biodegradation
- Cheap, good migration, higher ORP \rightarrow c-DCE
- Elimination of nitrates, sulfate, dissolved oxygen

Horice site (CZ)

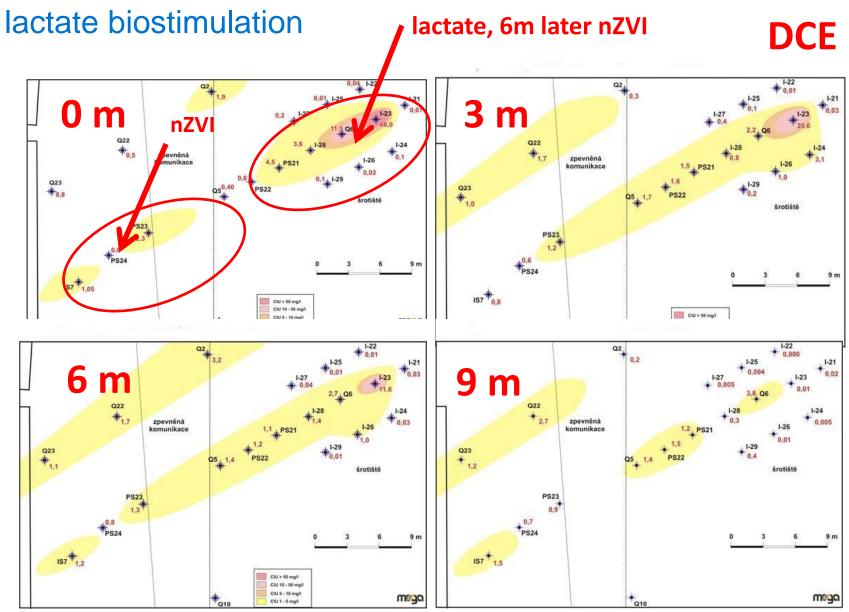
- Application of lactate
- 3x application nZVI
- Reduction by nZVI (48%)
- Combined reduction (76%)



Combination with bioremediation

lactate biostimulation PCE lactate, 6m later nZVI Q2 3,5 0,1 1-22 1-22 Q2 3,5 0,0021-25 1-21 **0** m 3 0.1 1-27 nZVI 022 1-24 0,008, 1-28 zpevněná zpevněná 0.007 6.8 + PS21 1-26 komunikace komunikace 4 _ PS21 0.003 0.5 1-29 Q23 0,5 0,017-29 Q23 **PS22** šrotiště šrotiště **PS23** PS23 21,0 PS24 11.0 PS24 9 m IS7 7 2 187 CIU > 50 mg1 CIU > 50 mg/l Q2 0,08 0,004 1-22 1-22 92 • 0,001 0,0021-25 1-25 1-21 •0,004 6 m 0.01 9 1-27 1-23 0,004, 1-27 €¹⁻²³ m 0.002 0.00 0.025 Q6 06 Q22 0,019 Q22 0,2 0,001, 1-28 1.28 1-24 1-24 zpevněná 0.00 0,001 zpevněná 0.00 1-26 0,007 PS21 1-26 komunikace 0,003PS21 komunikace 0.001 0.003 Q23 0,5 Q23 0,2 1-29 1.20 0.0005 PS22 šrotiště šrotiště PS23 PS23 11,6 PS24 9 m 6 IS7 6.6 CIU > 50 mg/l CIU 10 - 50 mg/ 0,05 0,7 CIU 5 - 10 mg/l m Q10 Citi 1 - 5 mol Q10

Combination with bioremediation



Combination with DC (EK-nZVI)

Principle of reaction

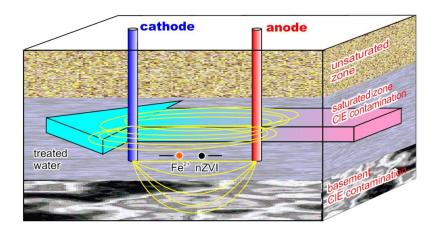
- Chemically supported reductive de-chlorination of CIE
- substitution of chlorine protons, while the electrons

$$CI_2C=CCI_2 + 4H^+ + 8e^- \longrightarrow H_2C=CH_2 + 4CI^-$$

- For the successful running of the reaction it is necessary to create a significant excess of protons and electrons in a geochemical system.
- By **Fe⁰** reaction with water.

 $5H_2O + Fe^0 \longrightarrow Fe^{3+}(OH)^{-}_{3} + H_2 + 2OH^{-} + 3H^{+} + 3e^{-}$

 Similarly by providing electrons using the DC electric field.



Combination with DC (EK-nZVI)

Principle of reaction



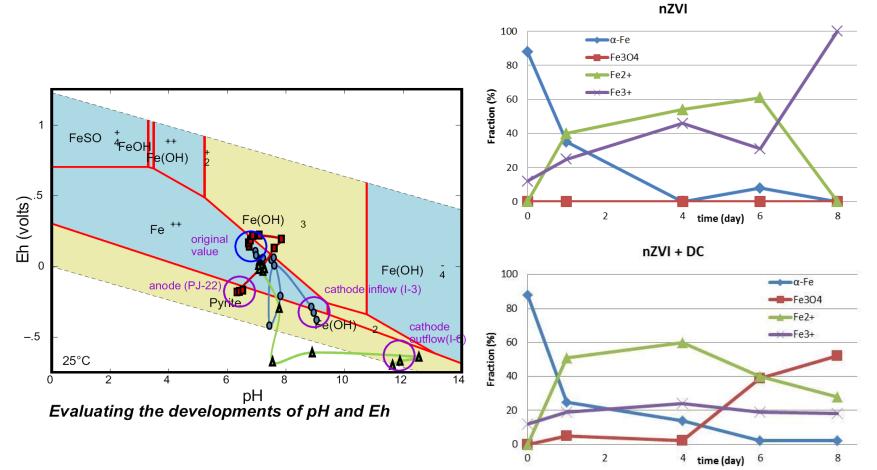
Cathode (RED) Water reduction pH increase O_2 reduction ORP decrease Anode (OX) Water oxidation pH decrease

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ORP increase

Combination with DC (EK-nZVI)

Final products of oxidation After 30 days 20% of Fe²⁺



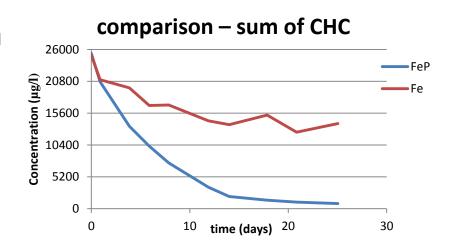
Combination with DC field

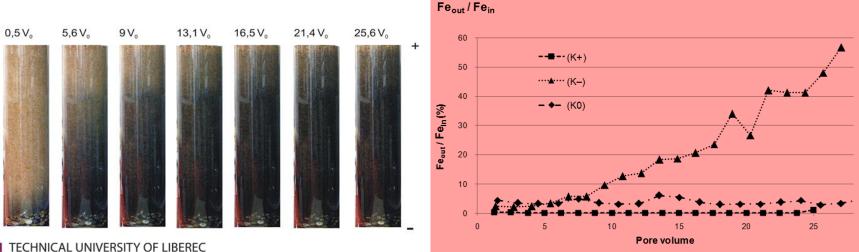
Laboratory experiments

- Principle in lab: DC ~1V/cm
- nZVI concentration 0.5 g/l
- Higher Fe²⁺ conc.
- Lower Eh

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- Better migration
- Higher reactivity





Health and Safety Considerations NANOREM project

- Health and Safety is an important issue
 - Exposure to NPs. should be considered for all "nano" products
 - Importance of studying and understanding of product behavior
 - Reduce risk by minimize contact of nanoparticles with persons
- Fate of nanoparticles in environment NANOREM
 - Possibility of controlled agglomeration by application of CaCl₂
- Current knowledge and future direction
 - Toxicity of nZVI towards water organisms
 - EU REACH legislation
- Two most recent studies including NANOFER product
 - Erik J. Joner et al. DDT degradation efficiency and ecotoxicological effects of two types of nanosized zero-valent iron (nZVI) in water and soil
 - Arturo A. Keller et al. *Toxicity of Nano-Zero Valent Iron to Freshwater and Marine Organisms*

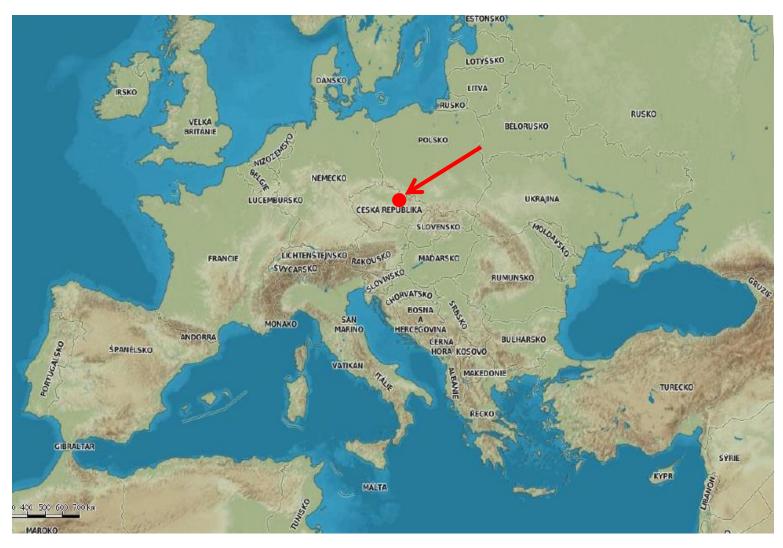
Innovative nZVI Manufacturing

Toxicity

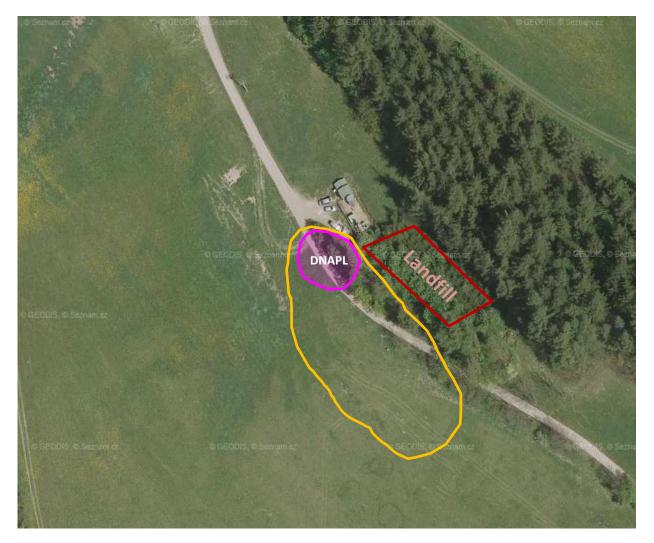
Soil Slurry: ecotoxicological effects and DDT degradation					
Organism	Endpoint	nZVI-B		NANOFER 25S	
		Aqueous p.	Solid p.	Aqueous p.	Solid p.
Earthworms	Mortality	n.a.	+	n.a.	0
	Growth	n.a.	-	n.a.	+
	Reproduction	n.a.	-	n.a.	+
Ostracods	Mortality	-	-	0	0
	Growth	-	-	0	+
Bacteria	Growth	-	n.a.	0	n.a.
Barley	Germination	+	-	0	-
	Root growth	-	-	+	+
Flax	Germination	-	-	0	0
	Root growth	-	-	+	0
DDT reduction in 24h		92.4%		78.3%	

ecotoxicological zero-valent iron effects of two types of nanosized degradation efficiency and (nZVI) in water and soil DDT

Pisecna – site view



Pisecna – site view



Pisecna site

- Former hazardous waste landfill
- Fractured bedrock area
- Chlorinated ethenes (Cl-E) and chlorinated ethanes (Cl-A) contamination
- Drinking water sources in the neighbourhood
- High reactivity needed for TCA degradation
- No special permits needed from water authority

Pisecna – Comparative lab-tests

Comparative test for 5 nZVI types:

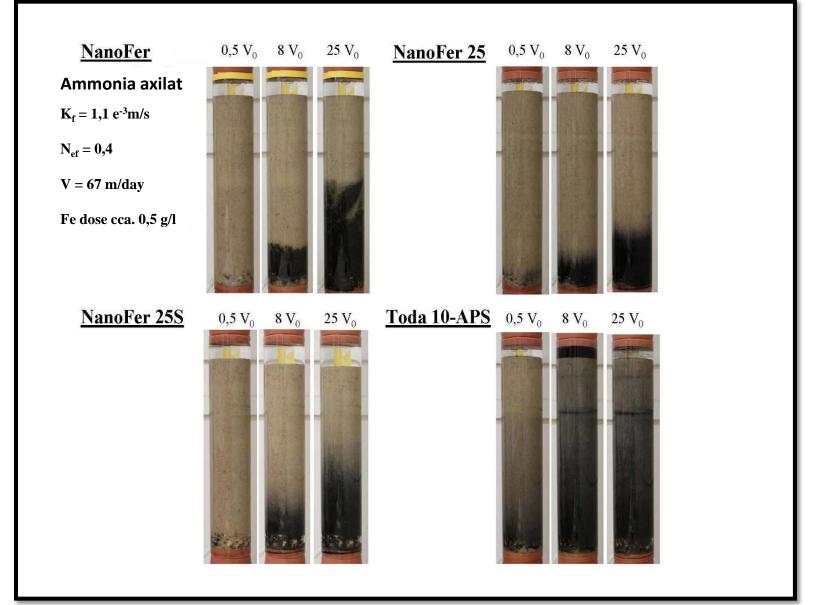
- borohydride nZVI (Zhang 2003)
- RNIP (Toda)
- NANOFER 25 without surfactant
- NANOFER 25S modified by TWEEN
- NANOFER ... modified by axilate
 - Axilate = Na salt of polyacrylic acid



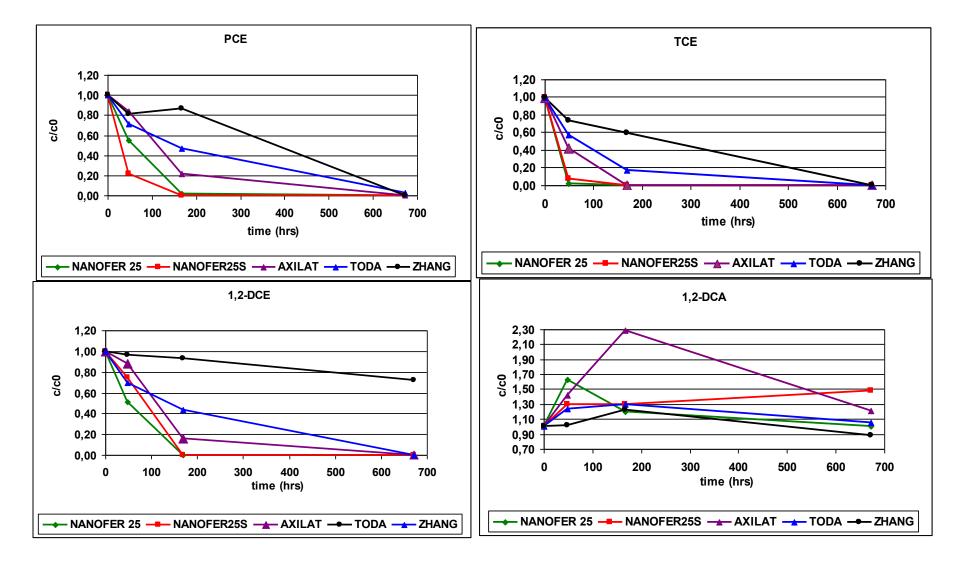
Tested properties:

- aggregation DLS
- sedimentation column tests
- mobility column tests
- reactivity kinetic tests, various nZVI concentration
 3 real ground water
 - 2 artifically mixed wat
 - 2 artifically mixed water

Pisecna – mobility tests

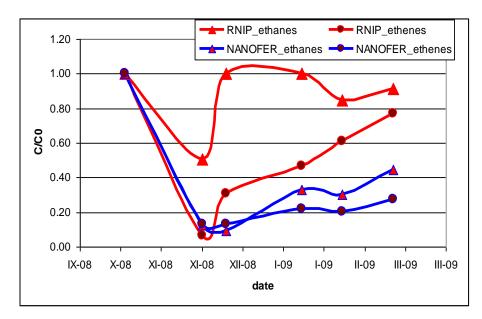


Pisecna – reactivity tests



Pisecna – pilot application

- RNIP vs. NANOFER25
- •Geological conditions not equal
- •CHC concentrations similar
- •Cl-Ethenes = both irons performed well
- •Cl-Ethanes = TODA iron worse (Δ)





The future of nZVI in Europe

State of the art and future developments

State-of-the-Art

- Different nZVI products available (dry, milled, slurry)
- Many lab and field tests accomplished –lectures to learn
- All points of view (reactivity, migration, storativity, transportability, toxicity,...)

Technical challenges:

- Successful field-scale applications in EU countries (needed for method acceptance/growth)
- Rigorous cost-effectiveness comparisons with other methods

Caveat



This project received funding from the European Union Seventh Framework Programme (FP7 / 2007-2013) under Grant Agreement No. 309517.

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Thank you for your attention

